## **CHART OF COLORANT BLENDS**

While it is possible to produce clay bodies that are quite attractive by the addition of single colorants, better hues and more subtle shades of color are possible by the addition of two or more colorants that will react favorably with one another.

The following guide is given to help the potter select combinations intelligently, for not all colorants react favorable with one another

ants rea	ct favorably with o	ne another.					
Chrome	+ cobalt oxide	= blue-green			rutile	=	ochre-browns
	copper oxide	= green					with mottling
	ilmenite	= warm green			vanadium	=	ochre-yellows
	iron oxide	<ul><li>grayed green</li></ul>					
	manganese	= brown (muddy)	Manganese	+	chrome oxide	=	browns (muddy)
	nickel oxide	= brown			cobalt oxide	=	blue-purples
	rutile	<ul><li>warm green</li></ul>			copper oxide	=	brown to black
	vanadium	= yellow-green			ilmenite	=	spotted browns
					iron oxide	=	browns
Cobalt	+ chrome oxide	= blue-green			nickel oxide	=	gray to brown
	copper oxide	= blue-green			rutile	=	mottled browns
	ilmenite	= mat gray-blue			vanadium	=	yellow-browns
	iron oxide	<ul><li>grayed blue</li></ul>					
	manganese oxide	= blue-purple	Nickel	+	chrome oxide	=	browns (greenish)
	nickel oxide	<ul><li>grayed blue</li></ul>			cobalt oxide	=	gray to blue
	rutile	<ul> <li>warmed grayed blue</li> </ul>			copper oxide	=	grayed greens
	vanadium	<ul> <li>grayed or ochre yellow</li> </ul>			ilmenite	=	browns
					iron oxide	=	browns (grayed)
Copper	+ chrome oxide	= green			manganese oxide	=	gray to brown
	cobalt oxide	blue-green			rutile	=	brown
	ilmenite	= warm gray-green			vanadium	=	gray to ochre-
	iron oxide	<ul> <li>warm grayed greens</li> </ul>					browns
	manganese oxide	<ul> <li>brown to metallic black</li> </ul>					
	nickel oxide	<ul><li>grayed greens</li></ul>	Rutile	+	chrome oxide	=	warm greens
	rutile	<ul><li>warm mat greens</li></ul>			cobalt oxide	=	warm mottled blues
	vanadium	<ul><li>warm yellow-greens</li></ul>			copper oxide	=	warm greens
					ilmenite	=	spotted browns
Ilmenite	+ chrome oxide	= warm green			iron oxide	=	ochre-browns
	cobalt oxide	= mat gray-blue			manganese oxide	=	mottled browns
	copper oxide	<ul><li>warm gray-green</li></ul>			nickel oxide	=	patterned browns
	iron oxide	= tan to spotted brown			vanadium	=	ochre-yellows
	manganese oxide	= brown to dark brown					
	nickel oxide	= brown	Vanadium	+	chrome oxide	=	yellow-greens
	rutile	<ul><li>spotted browns</li></ul>			cobalt oxide	=	ochre-yellows
	vanadium	= yellow-browns			copper oxide	=	warm yellow-greens
					ilmenite	=	yellow-browns
Iron	+ chrome oxide	= black-green			iron oxide	=	ochre-yellows
	cobalt oxide	<ul><li>grayed blues</li></ul>			manganese oxide	=	yellow-browns
	copper oxide	<ul> <li>warm green, metallic</li> </ul>			nickel oxide	=	gray to ochre-
		greens, and black					browns
	ilmenite	<ul><li>spotted browns</li></ul>			rutile	=	mottled ochre-
	manganese oxide	= browns					yellows
	nickel oxide	= grayed browns					